DE NOVO SYNTHESIS OF D-ALANINE IN GERMINATING PISUM SATIVUM SEEDLINGS

TADASHI OGAWA, YOSHIKO KAWASAKI and KEI SASAOKA
Department of Nutrition, School of Medicine, Tokushima University, Tokushima, Japan

(Revised received 10 February 1978)

Key Word Index—*Pisum sativum*; Leguminosae; pea; biosynthesis; D-alanine; γ-L-glutamyl-D-alanine; N-malonyl-D-alanine.

Abstract—The amounts of D-alanine derivatives, γ -L-glutamyl-D-alanine and N-malonyl-D-alanine, increase rapidly during the early growth of pea seeds. Pyruvate- $[1^{-14}C]$, L-alanine- $[U^{-14}C]$, D-alanine- $[U^{-14}C]$, L-alanine- $[U^{-14}C]$, L-alanine- $[U^{-14}C]$, L-alanine- $[U^{-14}C]$, L-alanine- $[U^{-14}C]$, D-alanine- $[U^{-14}C]$, L-alanine- $[U^{-14}C]$

INTRODUCTION

In previous studies [1-4], the authors identified a number of D-amino acids, such as D-alanine, D- α -amino-n-butyric acid, D-aspartic acid and D-glutamic acid in pea seedlings. Among them, D-alanine derivatives, γ -L-glutamyl-D-alanine (GA) and N-malonyl-D-alanine (MA), were found to be extensively produced during early germination [5]. D-alanine is the most widely reported D-amino acid in higher plants in the free or conjugated state [6]. However, its origin is still obscure, since it is not known whether it is actually synthesized de novo. For this reason the fate of labelled compounds fed to pea seedlings has been studied.

RESULTS AND DISCUSSION

Surface sterilized pea seeds were germinated for various periods of time in the dark to select the stage of growth at which the greatest accumulation of p-alanine derivatives occurs. The amounts of GA rapidly increased during the first 8 days of germination reaching a maximum, ca 3 µmol/seedling. MA also increased steadily during

Table 1. Utilization of ¹⁵NH₃ for p-alanine synthesis by pea seedlings

Amino acid	15N incorporated (atom % excess)
Free L-alanine	2.95
Free D-alanine	2.65
Conjugated D-alanine*	1.14

¹⁵NH₄Cl was supplied to the seedlings at the concn of 10 mM (99.5 at % excess) for 24 hr.

germination. The maximum production of D-alanine was observed between days 6 and 8 of seedling growth and was about 0.7 µmol/day/seedling. In contrast, free D-alanine was kept at low levels (ca 0.05 µmol/seedling) through the experimental periods. Therefore, the 6-day-old seedlings were used for the isotope experiments.

Pyruvate-[1-14C], L-alanine-[U-14C], D-alanine-[U-14C] and 15NH₄Cl were administered separately to the

Table 2. Utilization of ¹⁴C label for synthesis of D-alanine by pea seedlings

Compound administered*	Specific activity (μCi/μmol)	14C incorporated (dpm/μmol) Free Conjugated†			
		L-alanine	D-alanine	D-alanine	L-glutamic acid
Pyruvate-[1-14C]	14	2111	822	141	148
Pyruvate-[1-14C] + L-alanine	14	********		131	196
Pyruvate-[1-14C] + L-alanine Pyruvate-[1-14C] + D-alanine	14			55	133
L-Alanine-[U-14C]	168	*******		6552	4490
D-Alanine-[U-14C]	43	MANAGEN.		12776	783

^{*} Labelled compounds were supplied at the amounts of 0.5 µCi/seedling for 5 hr. Unlabelled compounds were added at the concn of 2 mM. † Degradation products derived from GA.

^{*} Degradation product obtained from GA.

Table 3. Utilization of L-alanine-[U- 14 C] and L-alanine- 15 N for D-alanine synthesis by pea seedlings. The two isotopically labelled compounds were combined to yield a soln which contained 28 μ mol of L-alanine per ml (7.14 \times 10⁵ dpm/ μ mol. 99.3 atom 6 excess) and then supplied to the seedlings for 24 hr

Amino acid	¹⁴ C incorporated (dpm/μmol)	Dilution	¹⁵ N incorporated (atom % excess)	Dilution
p-alanine*	1501	1:48	0.99	1 - 100

^{*} Degradation product derived from GA.

seedlings to see if D-alanine was formed from them. As shown in Table 1, germinating pea can utilize the exogeneous nitrogen to form D-alanine. The 15N label was effectively incorporated into free p-alanine as well as free L-alanine, while the apparent incorporation into the D-alanine moiety of GA was about 50 % of that found in free D-alanine. These data suggested that active synthesis of D-alanine and accumulation of GA proceeded in the 6-day-old seedlings. The utilization of radioactive carbon to form D-alanine is shown in Table 2. Pyruvate-[1-14C] was effectively incorporated into free and conjugated D-alanine. After Van Slyke decarboxylation with ninhydrin[6], 95 % of the ¹⁴C in p-alanine was recovered as CO₂, demonstrating that all the label was retained in C-1 of D-alanine. The incorporation of L-alanine-[U-14C] into D-alanine was much larger than that obtained with pyruvate-[1-14C]. The addition of unlabelled L-alanine to the medium of pyruvate-[1-14C] reduced the incorporation of 14C into the D-alanine moiety of GA, whereas the radioactivity of the Lglutamic acid part was not affected. Such isotope competition was more effectively observed when unlabelled D-alanine was added to the medium containing labelled pyruvate. This and the result obtained with p-alanine-[U-14C] indicated a direct incorporation of p-alanine into GA, after its formation from L-alanine. Additional evidence for the direct formation of D-alanine from Lalanine was obtained by supplying L-alanine-[U-14C] and L-alanine-[15N] simultaneously to the seedlings. As shown in Table 3, the retention of 15N isotope in the D-alanine moiety of GA amounted to ca 50% of 14C label. If the reaction proceeded through an intermediate. such as pyruvate, the D-alanine formed would possess a carbon skeleton derived largely from L-alanine, but would contain only a small amoung of L-alanine nitrogen. This, therefore, strongly suggested that the formation of D-alanine from L-alanine involved the retention of the 15N-amino group of the latter. In order to examine this further, enzyme analyses were carried out in vitro. An extract of the decotyledonized 6-day-old pea seedlings, free of endogenous low MW substances, catalyzed the direct conversion of L-alanine to its D-enantiomer. Thus a racemase occurs in germinating pea seedlings. In conclusion, both the tracer and the enzyme analyses indicate that there is de novo synthesis of D-alanine in pea via a racemase reaction. The occurrence of a similar system in a higher plant has been reported in tryptophan

metabolism in the cell cultures of the tobacco plant [8].

EXPERIMENTAL

Plant material. Seeds of pea (Pisum sativum L. cv Alaska) were sterilized with Osvan (0.01%) and germinated in the dark on moist filter paper at 25° as described previously [9]. 10 seedlings (decotyledonized, 6-day-old) were placed in a small vessel containing the aq. soln of the appropriate labelled compounds and were incubated for given periods under the same conditions.

Determination of amino acids. At the end of the incubation, the roots were washed thoroughly with H₂O and then the seedlings were homogenized with 75% EtOH and extracted at 4°. Determination of p-alanine and its conjugates in the extract was performed by methods described earlier [5]. Separation of the isomer of alanine was carried out by using a chromatographic technique (Aminex A-4, 0.9 × 70 cm column, 0.2 M Na-citrate buffer, pH 4.25 [1]) after a derivatization to the respective L-leucyl dipeptides according to the method of ref. [10]. Identity of amino acids was confirmed by an automatic amino acid analyzer, Yanagimoto LC-5S.

Labelled compounds and radioactive assay. The isotopes were obtained with following specifications: Na pyruvate-[1-14C] (14 mCi/mmol), D-alanine-[U-14C] (43 mCi/mmol), L-alanine-[U-14C] (168 mCi/mmol), L-alanine-15N (99.5 at. %) and 15NH₄Cl (99.5 at. %). The 14C radioactivity was determined by a liquid scinillation spectrometer, Aloka LCS-602 equipped with an automatic external standardization system. 15N analysis was performed at Japan Isotopic Center, the Institute of Physical and Chemical Research, Tokyo, Japan

REFERENCES

- Ogawa, T., Fukuda, M. and Sasaoka, K. (1973) Biochim. Biophys. Acta 297, 60.
- Fukuda, M., Ogawa, T. and Sasaoka, K. (1973) Biochim. Biophys. Acta 304, 363.
- Ogawa, T., Bando, N. and Sasaoka, K (1976) Agr Biol. Chem. 40, 1661
- Ogawa, T., Kimoto, M. and Sasaoka, K. (1977) Agr Biol Chem. 41, 1811.
- Fukuda, M., Tokumura, A., Ogawa, T. and Sasoaka, K. (1973) Phytochemistry 12, 2593.
- 6. Robinson, T (1976) Life Sci. 19, 1097
- Van Slyke, D. D., Macfadyen, D. A. and Hamilton, P (1941)
 J Biol. Chem. 141, 671.
- 8. Miura, G. A. and Mills, S. E. (1971) Plant Physiol. 47, 483.
- Sasaoka, K. (1958) Mem. Res. Inst. Food Sci., Kyoto Univ. 14, 42.
- Manning, J. M and Moore, S. (1968) J Biol. Chem 243, 5591.